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Q5: In neutralisation reactions between an acid and alkali. Hydrogen ions react with hydroxide ions to produce water. Represent this in an equation. (1 mark) Strong and weak acids Q6: Complete the sentences using the words in the boxes. A acid is ionised in aqueous solution. A weak acid is ionised in aqueous solution. (3 marks) 1 17 7 4

an acid (hydrochloric, sulfuric or nitric) of unknown concentration; sodium hydroxide solution of known concentration; a suitable indicator, for example methyl orange or phenolphthalein

Acidic solutions have pH values lower than 7. Acid-alkali indicators can show whether a solution is acidic, neutral (pH 7) or alkaline. The table shows the colours for two different common ...

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2. Circle the correct bold word in the sentences below. Acids taste sour/feel soapy. Some acids and alkalis are corrosive/correlated. A concentrated solution of an acid is more/less corrosive than a dilute solution. A concentrated solution has fewer/more acid particles per litre than a dilute solution.

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Science Home Learning Task Year 8 Acids and alkalis

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