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stored in bond. Kinetic Energy.

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ANSWERS: Make the following conversions: 444 cal to joules = 1.86×10^3 J. 1.8 kJ to joules = 1.8×10^3 J. 0.45 kJ to calories = 1.1×10^2 cal. Classify each of these processes as endothermic or exothermic: condensing steam - exo. burning alcohol - exo. evaporating alcohol - endo. baking a potato - endo The specific heat capacity for silver is 0.24 J/goC.

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Correct answer: Explanation : Using the equation , n being the number of moles, R being 8.314, and T being the temperature in Kelvin, we can find the kinetic energy of two moles of gas in a container.

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Correct answer: Explanation : Using the equation , n being the number of moles, R being 8.314, and T being the temperature in Kelvin, we can find the kinetic energy of two moles of gas in a container.

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