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### 10.1 The Mole: A Measurement of Matter

### Mole A Measurement Of Matter

### Chemical Quantities

### 10.1 The Mole: Measurement of Matter

The mole (symbol: mol) is the unit of measurement for amount of substance in the International System of Units (SI). A mole of a substance or a mole of particles is defined as exactly  $6.022\ 140\ 76 \times 10^{23}$  particles, which may be atoms, molecules, ions, or electrons. In short, for particles  $1\ \text{mol} = 6.022\ 140\ 76 \times 10^{23}$ . The definition was adopted in November 2018 as one of the seven SI base ...

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### What Is a Mole and Why Is It Used in Chemistry?

10.1: The Mole a Measurement of Matter Vocabulary: Define the following terms. Mole- Avogadro's Number- Representative Particle- Practice Problems: 1. What are three ways to measure the amount of something?

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