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chemical reactions that make up the Calvin cycle. The chemical reactions of the Calvin cycle use carbon dioxide (CO₂) and energy from ATP and NADPH to make sugars. * AcAdemic VocAbulAry channel a passage, or a space that can be moved through Light-independent reactions take place in the stroma. stroma (fluid) Interactive Reader 3

Interactive Reader Chemical Reactions Answer

chemical reaction. Oxygen and glucose are the reactants in the reaction shown above. • Products are the substances made by a chemical reaction. Carbon dioxide and water are the products of the above reaction. Bond Energy In order for reactants to change into products, the bonds of the reactants must break, and new bonds must form in the products. Breaking a bond requires energy. Bond energy is

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Interactive Reader 144 Chemical Reactions SECTION 2 Name Class Date Chemical Equations continued What Are Mole Ratios? Consider again the reaction between magnesium and oxygen to produce magnesium oxide. $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ The balanced equation tells you that two Mg atoms react with one O_2 molecule to produce two MgO molecules.

CHAPTER SECTION 12 Chemical Equations

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A change in color, odor, or texture may show that a chemical change has happened. Many chemical changes produce or absorb heat. An increase in temperature happens when a chemical change releases, or gives off, heat. Wood burning is a good example of a chemical change that gives off heat.

CHAPTER SECTION 3 Chemical Properties

chemical reactions that make up the Calvin cycle. The chemical reactions of the Calvin cycle use carbon dioxide (CO_2) and energy from ATP and NADPH to make sugars. * AcAdemic VocAbulAry channel a passage, or a space that can be moved through Light-independent reactions take place in the stroma. stroma (fluid) Interactive Reader 3

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Interactive Textbook 259 Chemical Reactions SECTION 4 Name Class Date Energy and Rates of Chemical Reactions continued How Fast Do Reactions Occur? A chemical reaction happens only if reactant particles collide. Even when they do collide, a reaction can occur only if there is enough energy to break chemical bonds.

CHAPTER 14 SECTION 4 Energy and Rates of Chemical Reactions

These reactions commonly happen between rock, water, carbon dioxide, oxygen, and acids or bases. Chemical weathering changes both the chemical composition of rock and how the rock looks. Acids are substances that form hydronium ions, H_3O^+ , in water. Hydroni-

um ions can cause bonds in minerals to break. Bases are substances that form hydroxide ...

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scientists design experi-ments to see
what makes plants grow larger or faster.