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Chemistry Semester 2: Final Exam The Final Exam in this class is the Michigan Secondary Credit Examination for Chemistry. Exam Grades will be based on achieving the Department of Education recommended passing rate of 70% (For Example - correctly answering 70% of the questions =

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Enzymes can catalyse reactions through a variety of mechanisms. First step involves binding of substrate to the active site of the enzyme and formation of substrate enzyme complex

### Chemistry - 2nd Semester Review for Final Exam

### Lynn, Mr. / 1st Semester AP Chemistry - 2020-21

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Description: This final exam review bundle includes a 16-page review packet of material usually covered in the second semester of chemistry. Topics include:-Types of chemical reactions (S, D, SR, DR, C)-Stoichiometry calculations (Including particles and volume conversions, and limiting reactants)-T

Second semester of a two-semester sequence for non-chemistry majors. Mastery of the second semester material requires comprehensive understanding and recall of the first semester material. Organic chemistry of conjugated  $\pi$  systems, aromatics, carbonyl compounds, amines, carboxylic acids and their derivatives, carbohydrates, lipids and proteins.

### Chemistry Second Semester Final Exam

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2nd Semester Chemistry-575 Final Exam Review Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. How many moles of tungsten atoms are in  $4.8 \times 10^{25}$  atoms of tungsten? a.  $8.0 \times 10^2$  moles c.  $1.3 \times 10^{-1}$  moles b.  $8.0 \times 10^1$  moles d.  $1.3 \times 10^{-2}$  moles \_\_\_\_ 2.

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The tests that are in trial testing in spring/early summer 2020: GC20S – 2020 General Chemistry Exam for the second semester of the standard two-term General Chemistry Sequence GC21C – 2021 General Chemistry Conceptual Exam full year exam GC21CF -2021 General Chemistry Conceptual Exam first term year exam

### Exams | ACS Exams

Chemistry - 2nd Semester Practice Test Thermochemistry (Ch.11): Multiple Choice 24) What is the amount of heat required to raise the temperature of 200.0 g of aluminum by  $10\Delta C$ ? cal (specific

heat of aluminum =  $0.21 \text{ J/g} \times \Delta C$ )

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Biochemistry (461 and 463), and biological information processing classes (465, 661, 662, and 674; grouped because they cover overlapping material).

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