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(T or F) The approximately tetrahedral arrangement of electron pairs in a water molecule allows the electrons to be as far from each other as possible

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Chapter 9 Chemical Calculations and Chemical Formulas

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Chapter 9 (.1-. 3) Section 9.1 (Naming Ions) Atoms are electrically neutral because there is the same number of protons (+) and electrons (-) Ions are atoms, or groups of atoms, with a charge (either positive or negative) o Made by gaining or losing electrons o An Anion is a negative ion that gained electrons Anions are usually nonmetals Written like O²⁻Cations are usually metals Written like Ca²⁺

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undergoes such formation is (9) Study Guide for Content Mastery p orbitals and its one s orbital, the orbitals that result are called (10) orbitals. An example of a molecule that has this type of orbital is (11) 52 Chemistry: Matter and Change Chapter 9

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