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Chapter 2 The Chemistry of Life Chapter Test B. Multiple Choice. Write the letter that best answers the question or completes the statement on the line provided. ____ 1. The three particles that make up an atom are a. protons, neutrons, and isotopes. b. neutrons, isotopes, and electrons. c. positives, negatives, and electrons.

Chemistry of Life NOTES Chapter 2.1 and 2.2

This accompanies the Mader text Human Anatomy & Physiology 8th edition.

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the amount of energy needed to increase the temperature of 1 gram of a substance by 1 degree celcius. --> substances with a high specific heat have to have more energy added to change their temperature. Water has one of the highest specific heats.

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(required for some necessary chemical reactions in life that cannot be done on their own) enzymes proteins that act as biological catalysts; speeds up chemical reactions that take place in cells

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Multiple Choice - Lawndale High School

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a long chain of carbon atoms with with one hydrogen atom bonded to each carbon atom, some with a double covalent bond which is easy to break - plant and fish oils - liquid at room temperature.

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A subatomic particle that has no charge and that is found in t... atom (2.1-35) (physics and chemistry) the smallest component of an element h... Nucleus (2.1-35) The center of an atom which contains protons and neutrons. The basic unit of matter Center of the atom, which contains the protons and neutrons.

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atom: electron: a negatively charged particle: element: a pure substance that consists entirely of one type of atom

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