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Radioactivity •Radioactivity is the process by which nuclei emit particles and rays as they break down. •The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

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Nuclear Chemistry Worksheet K Directions: Identify the following as alpha, beta, gamma, Name: Period: or neutron. Date: 231 90 91 Kr + 3 36 5. 6. 8. 9. Nuclear decay with no mass and no charge An electron Least penetrating nuclear decay Most damaging nuclear decay to the human body Nuclear decay that can be stopped by skin or paper. 3. Owl phcx ...

Unlike chemical reactions, nuclear reactions are not affected by changes in temperature, pressure, or the presence of catalysts. Also, nuclear reactions of a given radioisotope cannot be slowed down, sped up, or stopped. Three types of nuclear radiation are alpha radiation, beta radiation, and gamma radiation.

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Chemistry 1110 - Chapter 5 - Nuclear Chemistry - Practice Problems Page | 1 Chapter 5 - Nuclear Chemistry - Practice Problems 1. Fill in the missing information in the chart: 2. What is the nuclear symbol for a radioactive isotope of copper with a mass number of 60? A) Cu B) Cu C)  $^{29}\text{Cu}$  D) Cu E) Cu 3.

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cle: a fast-moving electron formed by the decomposition of a neutron ... number of protons + number of neutrons in the nucleus in nuclear chemistry ...

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Unlike chemical reactions, nuclear reactions are not affected by changes in temperature, pressure, or the presence of catalysts. Also, nuclear reactions of a given radioisotope cannot be slowed down, sped up, or stopped. Three types of nuclear radiation are alpha radiation, beta radiation, and gamma radiation.

### **Chapter 25**

Chapter 21 Nuclear Chemistry Chapter 21-Assignment A: Natural Radioactivity: Where Does It Come From? Where Does It Go? Most chemists study the results of electron sharing (covalent bonding, Chapters 10 and 11) and electron transfer (redox reactions, Chapter 19). Some chemists and physicists, however,

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tron ... number of protons + number of neutrons in the nucleus in nuclear chemistry ...

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