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1. a 2. b 3. c 4. c 5. a 6. b 7. c 8. Polarity in a water molecule is caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or basic. 10.

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Chapter 2: Introduction to the Chemistry of Life. The elements carbon, hydrogen, nitrogen, oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things. They form the carbohydrates, nucleic acids, proteins, and lipids (all of which will be defined later in this chapter) that are the fundamental molecular...

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Chapter 2: The Chemistry of Life Biology: McDougal Littel pages 34-64. Below you find the classroom assignments and PPT's used for Chapter 2, The Chemistry of Life. You may use this website for access to PPT's, guided notes, and make up assignments. The Chemistry of Life Assignments.

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Chapter 2: The Chemistry of Life Vocabulary. scale with values from 0-14, used to measure the concentration of H<sup>+</sup> ions in a solution; a pH of 0-7 is acidic, a pH of 7 is neutral, and a pH of 7-14 is basic.

Chapter 2: The Chemistry of Life. What do you see when you look at this picture? Is it just a mass of tangled ribbons? Look closely. It's actually a complex pattern of three-dimensional shapes. It represents the structure of a common chemical found inside living cells. The chemical is a protein called hemoglobin.

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Chapter 2: The Chemistry of Life. when molecules are close together, a slight attraction can develop between the oppositely charged regions of attraction (van der Waals forces)

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a long chain of carbon atoms with two hydrogen atoms bonded to each carbon atom with a single bond (except the carbon atom at the end which is bonded to three hydrogen atoms) - animal fats - solid at room temperature

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