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This is a basic look at elements and atomic structure.

BIOLOGY I. Chapter 2 – The Chemical Context of Life Life Has a Unique Chemistry • Chemists recognize 92 elements occurring in nature. About 25 are known to be essential to life, with 4 predominant elements: oxygen (O), carbon (C), hydrogen (H), and nitrogen (N) make up 96% of living matter. • Other important elements in living

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Chapter 2: The Chemical Context of Life . This chapter covers the basics that you may have learned in your chemistry class. Whether your teacher goes over this chapter, or assigns it for you do review on your own, the questions that follow should help you focus on the most important points. Concept 2.1 Matter consists of chemical elements in pure form and in combinations called compounds . 1.

Chapter #2 - The Chemical Context Of Life - Notes. 1. A Chemical Connection to Biology. 1. A single flowering tree species near the Amazon's headwaters in Peru. dominate tracts of forest. 2. The organism responsible for this pruning is the ant species. This organisms.

2: The Chemical Context of Life

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Chapter 2. The Chemical Context of Life Figure 2.1 Atoms are the building blocks of all the molecules found in the universe—air, soil, water, rocks . . . and also the cells of all living organisms. In this model of an organic molecule, the atoms of carbon (black), hydrogen (white), nitrogen (blue), oxygen (red), and sulfur (yellow) are shown in proportional atomic size.

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Chapter 2 The Chemical Context of Life Lecture Outline . Overview: Chemical Foundations of Biology. Living organisms and the world they live in are subject to the basic laws of physics and chemistry. Biology is a multidisciplinary science, drawing on insights from other sciences. Life can be organized into a hierarchy of structural levels.

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