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Haloarenes are less polar than haloalkanes and are insoluble in water. This is because of lack of hydrogen bonding. As a result, the attractive forces in haloarenes—water system remain less than the attractive forces in H₂O molecules which are hydrogen bonded.

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electronegative than carbon. So, the C-X bond is polarized with C having a partial positive charge and X having a partial negative charge. Preparation of haloalkanes:

Chemistry Notes for class 12 Chapter 10 Haloalkanes and Haloarenes. The replacement of hydrogen atom(s) in hydrocarbon, aliphatic or aromatic, by halogen atom(s) results in the formation of alkyl halide (haloalkane) and aryl halide (haloarene), respectively. Classification of Halogen Derivatives.

Chemistry Chapter-10: Haloalkanes and Haloarenes Question 10.1: Name the following halides according to IUPAC system and classify them as alkyl, allyl, benzyl (primary, secondary, tertiary), vinyl or aryl halides :

Haloalkanes are the halogen derivatives of hydrocarbons and these are of the following types depending upon the number of hydrogen atoms present in them. Haloarenes are aromatic compounds in which the halogen atom is attached directly to the carbon atom of the aromatic ring. The solutions of chapter 10 also give exemplary problems, HOTS (Higher Order Thinking Skills), MCQs exercises, assignments and worksheets which will help you to learn the concept comprehensively and to make haloalkanes ... Haloalkanes and Haloarenes Class 12 Chemistry Chapter 10 Notes 1. Haloalkanes are classified as fluoro, chloro, bromo or iodo compounds according to the type of halogen present and as mono-, di- tri-, tetra- haloalkanes, etc., according to the one, two, three, four, etc., halogen atoms respectively present in their molecule. 2.

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