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A) Change of colour in an acid and a base depends on the type of the indicator. B) Both acids and bases change colour of all indicators. C) None of the above D) If an indicator gives a colour change with an acid, it does not give a change with a base. E) If an indicator changes colour with a base, it does not change colour with an acid.

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Which acid is present in orange? Lactic acid. Citric acid. Methanoic acid. Oxalic acid. Correct Option is : 2. Solution : Orange contains citric acid. Question 6: In which of the following bleaching powder is used? Oxidising agent in chemical industry. For disinfecting drinking water. For bleaching wood pulp in paper factories. All of these ...

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Acetic acid . Sulphuric acid . Sodium Hydroxide . Correct Option is : 3. Solution : Question 5: Choose the correct option. In neutralization reaction . Acid and base react. Only water and heat are produced. Heat is not evolved. None of the statements are correct. Correct Option is : 1. Solution : In a neutralization reaction, acid and base ...

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Acids and bases interact with each other in what is called a neutralization reaction. The products of the reaction are a salt and water. The pH is "neutralized" to 7 if both the acid and base fully react.

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Acids Bases Salts Pre Test Acids and bases can cause. irritation to skin. corrosion. both 1 and 2. none of the above. Correct Option is : 3. Solution : Acids and bases should be handled with great care as they are corrosive in nature, cause skin irritation and are harmful to skin. Question 7: Limewater is. CaO. Ca(OH) 2. CaCO 3. Mg(OH) 2 ...

Acids, Bases and Salts 21 Activity 2.6 n Take about 2 mL of dilute NaOH solution in a test tube and add two drops of phenolphthalein solution. n What is the colour of the solution? n Add dilute HCl solution to the above solution drop by drop. n Is there any colour change for the reaction mixture? n Why did the colour of phenolphthalein change after the addition

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Mineral Acids: Mineral acids are acids prepared from minerals. For example, Hydrochloric acid (HCl), Sulphuric Acid (H₂SO₄), and nitric acid (HNO₃) etc. Also Check ⇒ Dilute Acids. Bases. The most common characteristic of bases is their bitter taste and soapy feel. A base is a substance that renders hydroxyl ion(OH⁻) in their aqueous ...

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CHAPTER2 Acids, Bases and Salts

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Ch 2 Acid, Bases and Salt MCQ Test 3| Class 10th ...

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Acids Bases and Salts ICSE Class-10 Concise Chemistry ...

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Quiz: Take This Important Acid, Base And Salt Test ...

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