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This reaction is readily reversible. If carbonic acid were added to this system, some of it would be converted to bicarbonate and hydrogen ions, as the chemical system seeks equilibrium. (6.2.2) $\text{H}_2\text{CO}_3 \rightleftharpoons \text{H}^+ + \text{HCO}_3^-$. The examples above examine "idealized" chemical systems as they might occur in a test tube.

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